

Cambridge International Examinations Cambridge International General Certificate of Secondary Education

## PHYSICAL SCIENCE

0652/61 October/November 2016

Paper 6 Alternative to Practical MARK SCHEME Maximum Mark: 60

Published

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Page 2	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
1(a)(i)	pestle and mortar ;	1
1(a)(ii)	bubbles will stop ;	1
1(a)(iii)	draw filter funnel and receptacle ; complete piece of filter paper ; filtrate and residue labelled correctly ;	3
1(a)(iv)	heat/boil ; saturate/remove some of water/crystalisation point/partly evaporate ; cool/leave ;	3
1(b)(i)	heat (until white) ;	1
1(b)(ii)	blue ;	1
	Total	10

Question	Answers	Marks
2(a)(i)	add sodium hydroxide (solution)/NaOH ; green ppt ;	2
2(a)(ii)	add dilute nitric acid/HNO <sub>3</sub> ;	3
	then add barium nitrate solution / Ba(NO <sub>3</sub> ) <sub>2</sub> ; white ppt. ;	
2(b)(i)	hydrogen/H <sub>2</sub> ;	1
2(b)(ii)	white ppt ; ppt dissolves/becomes colourless solution/soluble in excess ;	2

Page 3	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
2(c)(i)	displacement/redox/cation reduced/it is reduced/Fe <sup>2+</sup> goes to Fe/it is replaced by Mg/it is replaced by Mg <sup>2+</sup> ;	1
2(c)(ii)	exothermic ;	1
	Total	10

Question	Answers	Marks
3(a)(i)	6.5 ;	1
3(a)(ii)	65 ;	1
3(a)(iii)	Appropriate precaution (either written or shown on diagram) ; e.g. take reading at eye level/use of set square to ensure rule vertical /use of fiducial aid	max 1
3(b)	31. <u>0</u> ;	1
3(c)	T = 1.55; $T^2 = 2.4$ ;	2
3(d)	Suitable choice of scales (more than half the grid used) ; At least 4 plots correct to ½ small square ; Good best-fit straight line with a ruler, omission of anomalous point ;	3
3(e)	Yes agree (no mark) (straight) line through the origin	max 1
	No disagree (No mark) all points/anomaly not on the (straight) line	
	Total	10

Page 4	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
4(a)(i)	variable resistor ;	1
4(a)(ii)	to mix ice and steam ;	1
4(a)(iii)	all water/all liquid/no ice ;	1
4(b)	260 ; 260 × 5 × 24 = 31 200 J ;	2
4(c)(i)	113 ;	1
4(c)(ii)	(13g of) condensed / liquefied steam ;	1
4(di	Any 2 steam condensing / cooling <u>in the tube</u> / on the way to the ice ; not all steam heats the ice ; ice takes in heat from the environment ;	2
4(d)(i)	insulation/lid;	1
	Total	10

Page 5	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
5(a)(i)	oxygen/O <sub>2</sub> ;	1
5(a)(ii)	Hydrogen/H <sub>2</sub> ;	1
5(b)(i)	litmus <b>OR</b> UI ;	1
5(b)(ii)	gas will not change the colour of red <b>and</b> blue litmus/ UI <b>and</b> green or pH 7;	1
5(c)	diagram showing the inverted <b>test</b> -tube with the open end under water ; water risen into the test-tube ;	2
5(d)	gas <b>V</b> = ammonia/NH <sub>3</sub> ; gas <b>W</b> = hydrogen chloride/HCI/sulfur dioxide/SO <sub>2</sub> ;	2
5(e)	add limewater to test-tube and shake ; (limewater goes) white precipitate/milky ;	2
	Total	10

Page 6	Mark Scheme	Syllabus	Paper
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Question	Answers	Marks
6(a)	A in series with the power source <b>AND</b> V in parallel ;	1
6(b)	0.65(A); 1.5(V);	2
6(c)	wire L = 1.5 ; wire M (= 1.5/0.65 =) 2.3 ; ohms/Ω ;	3
6(d)	minimum of 3 lengths ; minimum 10cm range ; control <b>ONE</b> from material/cross-section/temperature ; graph of resistance against length ;	4
	Total	10