

Cambridge IGCSE™

CO-ORDINATED SCIENCES

0654/52

Paper 5 Practical Test

October/November 2020

MARK SCHEME

Maximum Mark: 60

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2020 series for most Cambridge IGCSE™, Cambridge International A and AS Level and Cambridge Pre-U components, and some Cambridge O Level components.

This document consists of **10** printed pages.

Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.

2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.

3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).

4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.

5 'List rule' guidance

For questions that require *n* responses (e.g. State **two** reasons ...):

- The response should be read as continuous prose, even when numbered answer spaces are provided.
- Any response marked *ignore* in the mark scheme should not count towards *n*.
- Incorrect responses should not be awarded credit but will still count towards *n*.
- Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
- Non-contradictory responses after the first *n* responses may be ignored even if they include incorrect science.

6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

| Question | Answer | Marks |
|----------|--|----------|
| 1(a)(i) | blue recorded for 0.0; full set of results; yellow / green recorded for 0.5; red recorded for 2.0; colour recorded for U ; | 5 |
| 1(a)(ii) | correct estimate and explanation from Table 1.1; | 1 |
| 1(b) | Benedict's ; | 1 |
| 1(c) | 1 from: tongs / glove AND protect hands / skin AND from burns / hot water ; goggles AND protect eyes AND from corrosive / liquids / Benedict's / solution / chemical / substance / hot water ; water bath AND protect from fire / burns AND flammable ; | 1 |
| 1(d)(i) | prevent contamination AW; | 1 |
| 1(d)(ii) | use colour chart / more intermediate glucose concentrations / examples ; | 1 |
| 1(e)(i) | ethanol / alcohol and water ; | 1 |
| 1(e)(ii) | White emulsion ; | 1 |
| 1(f) | ethanol / alcohol <u>and</u> flammable; | 1 |

| Question | Answer | Marks |
|----------|--|-------|
| 2(a) | <p>1 mark from each section and any 2 others If one section is missed max 6 etc.</p> <p>apparatus ruler; temperature changing apparatus ; method of illumination; soil / cotton substrate;</p> <p>method seedling in pots; at least two temperatures measured / hot and cold ; water them (not just water at the strat);</p> <p>measurement (and accuracy) measure height / length / size of shoot; after at least 5 days ; repeats / lots of seedlings; 4 or more different temperatures ;</p> <p>control variables same amount of water / specified amount ; same amount of light; same amount of nutrients / soil ; same amount of carbon dioxide; same species of plant;</p> <p>Process results and draw a conclusion calculate a mean / average ; plot graph of temperature against height ; see if higher / lower temperature increases / decreases the heights ;</p> | 7 |

| Question | Answer | Marks |
|-----------|---|-------|
| 3(a)(i) | fizzes / colourless solution ; | 1 |
| 3(a)(ii) | carbon dioxide AND (bubble through) limewater goes milky ; | 1 |
| 3(a)(iii) | carbonate (ions) / CO_3^{2-} ; | 1 |
| 3(b)(i) | ammonia AND turns (moist) red litmus paper blue ; | 1 |
| 3(b)(ii) | ammonium (ions) / NH_4^+ ; | 1 |
| 3(b)(iii) | wear goggles AND sodium hydroxide is corrosive ; OR point test-tube away from people ; AND if liquid spits out it will not become in contact with someone ; OR hold test-tube in a holder AND so you do not burn yourself holding with your hand ; | 1 |
| 3(c)(i) | white precipitate ; soluble / dissolves / colourless solution or ... white precipitate ; ... soluble / dissolves / colourless solution ; | 3 |
| 3(c)(ii) | to remove any soluble material ; | 1 |
| 3(c)(iii) | zinc ions / Zn^{2+} ; | 1 |

| Question | Answer | Marks |
|----------|---|-------|
| 4(a) | temperature and lowest temperature for 10 mm of magnesium ; all ten temperatures recorded ; all temperatures quoted to nearest 0.5 °C ; the temperature rise increases as the length of ribbon increases ; | 4 |
| 4(b) | all temperature rises correctly calculated ; | 1 |
| 4(c) | as length increases the temperature rise increases ; | 1 |
| 4(d) | Any one from: concentration of acid ; volume of acid ; thickness of magnesium ribbon ; | 1 |
| 4(e)(i) | loss of energy (to surroundings) ; | 1 |
| 4(e)(ii) | insulate beaker / use a plastic beaker or cup / use a lid ; | 1 |

| Question | Answer | Marks |
|----------|---|-------|
| 5(a) | l_0 recorded to the nearest 0.1 cm ; | 1 |
| 5(b) | l_1 recorded and $> l_0$ and e correctly calculated ; | 1 |
| 5(c) | correct calculation of k ; | 1 |
| 5(d)(i) | t recorded in Table 5.1 ; | 1 |
| 5(d)(ii) | t recorded in Table 5.1 and $>$ value in (i) ; | 1 |
| 5(e)(i) | both values of T correct ; | 1 |
| 5(e)(ii) | both values of T^2 correct ; | 1 |
| 5(f) | $k = 0.25 \pm 0.05$ (N / cm) ; | 1 |
| 5(g) | (expect YES) values are (very) close / not (very) far apart / within 10% of each other / the same ; | 1 |
| 5(h) | view perpendicularly to scale / rule close to spring / use a fiducial aid ; | 1 |

| Question | Answer | Marks |
|----------|---|-------|
| 6(a) | temperature recorded at $t = 0$ to nearest $0.5\text{ }^{\circ}\text{C}$; | 1 |
| 6(b) | all temperatures recorded <u>and</u> all times correct ; temperatures decreasing ; | 2 |
| 6(c)(i) | axes labelled with units ; suitable scale (\geq half the grid used) ; minimum of 4 plots correct $\pm \frac{1}{2}$ small square ; | 3 |
| 6(c)(ii) | best-fit curve drawn within $\frac{1}{2}$ square of points ; | 1 |
| 6(d) | answer correct from candidates' results ; | 1 |
| 6(e)(i) | gloves AND protect hand AND from burns / hot water OR goggles AND to protect eyes AND from hot water / burns OR pour slowly / no splashes when pouring AND protect hands / skin AND from burns / hot water OR tongs AND protect hand AND from burns / hot water | 1 |
| 6(e)(ii) | use a lid (between readings) / thicker beaker / insulate the beaker / use a plastic beaker ; | 1 |